

5 **TITLE OF INVENTION**

*Chen* A Biocompatible Aqueous Solution For Use in Continuous Ambulatory Peritoneal Dialysis.

10 **SUMMARY OF THE INVENTION**

Continuous ambulatory peritoneal dialysis (CAPD) is used to treat end stage renal failure (ESRF) by introducing an osmotically active solution into the peritoneal cavity. Toxic waste products and excess fluid move from the blood into the dialysate solution by diffusion and ultrafiltration across the peritoneum. Osmotic ultrafiltration occurs as a result of the addition of hypertonic concentration of glucose to the dialysing solution. Due to the osmotic gradient between the blood and the CAPD solution the glucose draws water from the blood stream into the peritoneal cavity. The osmotic effect is transient and diminishes as the glucose is absorbed and/or metabolised.

In CAPD the dialysis solution is infused from collapsible plastic bags into the peritoneal cavity where it is retained for a period of time (referred to as the dwell time), after which it is drained and discarded. Generally, 3–5 treatments or exchanges of 1–3 litres each of CAPD solution are carried out daily, with an overnight dwell. The glucose concentration varies between 1.5 and 5% (w/v), with commercial CAPD solutions containing 1.5%, 2.5 or 4.5% glucose, with a high lactate content and various electrolytes which are present in more or less pH physiologic concentrations. CAPD patients also lose 5–10 grams of protein into the dialysate per day. Commercial CAPD solutions typically have an osmolarity of 300–700 mOsm/L, preferably 350–450 mOsmol/L, as taught by Patent 5,011,826.

Although peritoneal dialysis has some advantages over hemodialysis, including a substantial cost saving, there are several potential complications to CAPD. These include protein loss through the relatively highly permeable peritoneal membrane, absorption and metabolism of the added glucose resulting in weight gain and hyperlipidemia, which is particularly problematic in diabetic patients, who have a high incidence of ESRF (Ong– Ajyooth, L., Transp Proc 26: 2077, 1994).

An average patient absorbs about 150 grams of glucose from the dialysate per day, which for many patients is an excessive source of carbohydrate and results in hyperinsulinemia and hypertriglyceridemia in non–diabetic patients,

5 which contributes to atherosclerotic disease. This series of events likely contributes to cardiovascular disease which is the most common cause of death among patients with ESRF.

10 Chronic exposure of the peritoneal membrane to the hypertonic and acidic CAPD solution (pH 5–6.2) can result in a loss of its function as an ultrafiltration membrane, leading to increased permeability of the peritoneal membrane and an increased rate of absorption of glucose from the dialysis solution and a loss of ultrafiltration capability. (Breborowicz et al *Advances in Peritoneal Dialysis* 8: 11, 1192 and Breborowicz et al *Nephron* 67: 350, 1994).  
15 Peritoneal biopsy samples from patients chronically dialysed with CAPD solutions show a typical epithelial reaction to irritation, mesothelial cell proliferation, as well as a decrease in the number of microvilli which normally line the mesothelial cell surface (Dobbie, J.W., Lloyd, J.K., Gall, C.A. In R. Khamma, K.D. et al Eds. *Advances in peritoneal dialysis*. Toronto. U of Toronto Press, 3, 1990: Friedlander, M. J *Lab Clin Med* 122: 639, 1993). A chronic inflammation of the peritoneum is also a consequence of chronic CAPD treatment, possibly related to the acidic nature of the CAPD solution (Lewis, S. & Holmes, C. *Periton Dial Int* 11: 14, 1991; Beelen, R.H.J. et al In Maher J.F., Winchester, J.F. Eds. *Frontiers in peritoneal dialysis*. New York: Field, Richj and Associates, 524, 1986; Bos, H.J. et al *Nephron* 20 59: 508, 1991), and which leads to healing (Weiczorowska, K. et al *Short Reports?*). Morphologic changes in the peritoneal structure also occur with chronic CAPD therapy, including fibrosis of the peritoneum (Chaimovitz, C., *Kidney Int* 45: 1226, 1994). Further, the use of the current relatively acidic and glucose hypertonic CAPD solutions results in a decrease in the function of peritoneal macrophages, again indicating a need for more physiologic and biocompatible CAPD solutions (deFijter, C.W.H. et al *Clin Nephrology* 39: 75, 1993).  
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35 As well, it has been shown that there is a loss of glycosoaminoglycans (GAG's) from the peritoneal membrane which results in a loss of filtration efficiency. It has been suggested that the loss of GAG's from the peritoneal membrane is a result of the increased production of free radicals by activated peritoneal leucocytes (Breborowicz, A. et al *Periton Dial Int* 11(Suppl): 35a, 1991) or because of a destructive action on interstitial tissue proteins (Fligiel, S.E.G. et al *Amer J Pathol* 115: 418, 1984). Supplementation of the dialysis fluid  
40 with the GAG chondroitin sulphate increases net ultrafiltration due to slower absorption of glucose and fluid from the peritoneal cavity (*Advances in Peritoneal Dialysis* 8: 11, 1992; *Nephron* 67: 346, 1994), possibly due to its ability to scavenge free radicals. Other GAG's, such as heparin and dermatan have also been reported

5 to scavenge free radicals (Hiebert, L., Liu, J.M., Semin Thromb Hemost 17: 42, 1991; Fracasso, A. et al J Amer Soc Neph 5: 75p, 1994). It has also been reported that hyaluronan (formerly known as hyaluronic acid), which also scavenges free radicals, protects the peritoneum from injury resulting from CAPD treatment (Wieczorowska, K. et al Perit. Dial. Int. 15:81, 1995). Supporting this is the finding  
10 that the dialysis fluid collected overnight has a higher concentration of hyaluronan than serum. For example, Yung, S. et al (Kidney Int 46: 527, 1994) found that hyaluronan levels increased in the dialysate from ESRF patients with or without peritonitis undergoing CAPD treatment, and that the peritoneal mesothelial cells were the likely source of the hylauronan. Hyaluronan is important in the regulation  
15 of cell proliferation during healing. Hyaluronan is a polymer of repeating molecules of N-acetylglucosamine and glucuronic acid; dermatan is composed of repeating units of N-acetylglucosamine and iduronic acid, and chondroitin is made up of glucuronic acid and N-acetylgalactosamine.

20 Breborowicz and Oreopulos have submitted a PCT patent application (EP-555087-A1) (priority 92US-830721) for the addition of free radical scavengers such as GAG's, including hyaluronic acid degradation products, to CAPD solutions during episodes of peritonitis to prevent against peritonitis-associated inflammatory reactions .

25 As noted above, N-acetylglucosamine (NAG) is a component of many GAG's. NAG is formed in almost all cells from glucose through a series of biochemical reactions which include the addition of the amine group from glutamine to glucose to form glucosamine, with N-acetylglucosamine being  
30 synthesized by way of acetyl-CoA. NAG then is converted to NAG-6-phosphate (which is converted into the epimer of NAG, N-acetyl-mannosamine 6-phosphate which is converted to N-acetylneuraminic acid 9-phosphate which is incorporated into sialic acids, gangliosides and glycoproteins ), to NAG-1-phosphate (which is converted into UDP-N-acetylglucosamine (UDP-NAG) which is incorporated into  
35 GAG's such as chondroitins and glycoproteins). The UDP-NAG is also converted into GAG's such as hyaluronan and glycoproteins. Thus, NAG is the primary building block of many essential tissue components, whether they are comprised of NAG itself or related amino sugars such as N-acetylmannosamine and N-acetylgalactosamine.

40 It has been shown that orally administered glucosamine and N-acetylglucosamine (NAG) are absorbed and distributed throughout the body rapidly, and incorporated into tissues and presumably into the GAG's of the body.

5 These compounds are incorporated into the GAG's of the peritoneal membrane to  
prevent their depletion thus maintaining the integrity of the peritoneal membrane,  
and preventing or at least slowing down, the loss of membrane function as an  
ultrafiltration membrane. Thus, the replacement of part or all of the glucose in the  
10 presently available CAPD solutions with amino sugars, especially NAG, should  
provide a more biocompatible peritoneal dialysis solution, while providing the  
necessary osmotic effect required for the removal of excess water and also removal  
of waste substances by solvent drag from patients with ESRF undergoing CAPD  
treatment. Unlike glucose, which is utilized by almost all microorganisms as a  
15 source of energy, the amino sugars are relatively less metabolized and not as likely  
to support microbial growth thus reducing the tendency for patients undergoing  
chronic CAPD treatment to develop peritonitis, a common and serious adverse  
event associated with CAPD treatment. Because of the rapid removal; of NAG and  
other amino sugars from the systemic circulation by way of their incorporation into  
GAG's and various amino sugar containing tissue components the extent of  
20 metabolism into lipids is significantly reduced, thus reducing the risk of obesity,  
protein malnutrition, dyslipidemia and hypertriglyceridemia, hyperinsulinemia etc  
and the related adverse metabolic consequences.

In order for NAG and related amino sugars to be useful as osmotic  
agents in CAPD solutions they must have a high chemical purity similar to that  
25 which would be required for use in pharmaceutical products, which means a  
minimum purity of 98.5%. NAG which is of this purity can be manufactured by two  
methods. The first is the acid digestion of crude chitin, which is a linear polymer of  
repeating units of NAG obtained from crab and shrimp shells and other  
crustaceans, followed by isolation of the deacetylation of the individual NAG units  
30 to glucosamine. The glucosamine is isolated and crystallized to a high level of  
purity and then is reacylated using acetic anhydride to N-acetylglucosamine,  
which is precipitated and recrystallized from alcohol, such that its purity is greater  
than 98.5%. The second method of manufacturing NAG, and the preferred method,  
is to obtain NAG from dried crustacean shell or crude chitin by direct enzymatic  
35 digestion with an ensemble of enzymes including chitinase and chitobiase, which  
degrades the chitin polymer of NAG into disaccharide units of chitobiase and then  
into monomer units of NAG directly, without having to undergo any organic  
synthetic step. The NAG is recrystallized from alcohol to a high degree of purity  
from ethanol. The enzymes required for this process are secreted into the growth  
40 media of various microorganisms, especially *Serratia marcescens*. Thus this  
method of manufacture not only provides NAG of a suitable purity for use in CAPD  
solutions but also permits the relatively inexpensive production of NAG as the chitin

5 or crustacean shells can be added directly to the cell-free growth medium from a culture of *S. marcescens* and the NAG readily isolated from the medium after a suitable reaction period. By varying the length of the enzymatic reaction time the production of polymers of varying units of NAG can be produced, which can be further refined and isolated as specific molecular weight entities by way of  
10 separation using available chromatographic techniques, and which can be isolated, crystallized and further purified by recrystallization using methods familiar to those skilled in the methods of carbohydrate chemistry isolation and purification.

Patent 5, 011,826 teaches that CAPD solutions can use galactose  
15 alone or with glucose in varying ratios as the osmotically active agents, whereas Patent 4,879,280 teaches that disaccharides such as lactose, saccharose, cellobiose etc can be used similarly, both together with suitable electrolyte additives. As well patent 4,879,280 also shows the use of trisaccharides, oligosaccharides and polysaccharides of a molecular weight less than 400,000  
20 such as raffinose, starch, inulin, pectin, dextrans, hydroxy-ethyl starch (HES) and the like. For example, colloidal polymers of glucose of 4-250 glucose units long and with an weight average molecular weight of about 16,200 and a number average molecular weight of 5,800 has been clinically evaluated as component of a CAPD solution (Kidney Int 46: 496, 1994: US Patent 4,886,789). The osmolality  
25 of a 7.5% solution of this glucose polymer, called Icodextrin, was 282 mOsm/kg and had a pH of 5.3. However, neither the available scientific literature nor the available patents teach the use of polymers or oligimers of amino sugars such as N-acetylglucosamine, N-acetylmannosamine or N-acetylgalactosamine and the like as the osmotically active components of CAPD solutions, which are the subject  
30 of the present invention.

Since the effectiveness of intraperitoneal dialysis depends on the presence of a hypertonic solution and osmolarity depends on the number of molecules in solution, large molecules such as GAG's provide little of value to the  
35 osmotic effect of the CADP solution, and the dialysis solution must still contain excess glucose. Since N-acetylglucosamine and related amino sugars, as well as the other sugar and/or acidic carbohydrates making up the GAG's have molecular weights similar to that of glucose, they would be osmotically active. Therefore, the inclusion of amino sugars, particularly N-acetylglucosamine, in a CAPD solution at  
40 concentrations ranging from 0.5 to 5%, with or without the presence of glucose, will provide an effective dialysis solution while being more biocompatible with the peritoneal membrane and thus preventing or slowing down the morphologic and functional deterioration of the peritoneal membrane and extending the time over

5 which ESRF patients may effectively use CAPD treatment. This provides several  
benefits, including substantial cost saving to the health care system by reducing the  
need for expensive hemodialysis, a lower rate of peritoneal infection for patients  
receiving CAPD treatment, a lesser risk of cardiovascular disease due to a  
10 reduction in the lipid changes typical of use of currently available CAPD solutions,  
and a better quality of life for such patients.

Currently marketed CAPD solutions have the following typical  
composition per 100 mL of solution. Dextrose anhydrous 1.5, 2.5 or 4.25 plus  
Sodium Chloride 567 mg, Sodium lactate 392 mg, Calcium Chloride dihydrate  
15 23.9 mg and Magnesium Chloride hexahydrate 15.2 mg. On a milliequivalence  
basis this represents 132 mEq Na/L, 3.24 mEq Ca/L, 1.5 mEq Mg/L, 101.75 mEq  
Cl/L and 36 mEq lactate/L. Alternately, the solution may contain malate, acetate or  
succinate in place of lactate. The solution typically has an osmotic pressure of 347  
mOsmol/L.

20 The CAPD solution of this invention is intended to provide similar  
electrolyte levels as currently available CAPD solutions, except that the osmotically  
active carbohydrate composition is different, being composed of acetylated and  
deacetylated amino sugars including N-acetylglucosamine, glucosamine,  
25 N-acetylgalactosamine, galactosamine, N-acetylmannosamine, mannosamine  
each alone, or in combination at varying concentrations or with varying  
concentrations of glucose, or oligomers of N-acetylglucosamine,  
N-acetylmannosamine, N-galactosamine, galactosamine, mannosamine, and  
glucosamine such that they are comprised of at least 2 carbohydrate units and not  
30 more than 12 units. The composition may be a mixture of oligimers of varying  
amounts of each oligimer either alone or in combination with each other. As well  
the CAPD solutions of this patent may contain additional osmotically active agents  
in varying proportions to the acetylated and deacetylated amino sugars such acidic  
carbohydrates which are also incorporated into the tissue glycosoaminoglycans  
35 (GAG's) such as glucuronic acid and iduronic acid.

In animal models of inflammatory bowel disease the colon becomes  
fibrotic, as does the peritoneum as a result of chronic intraperitoneal dialysis. The  
administration of a solution of NAG into the bowel of rats in which a chemically  
40 induced inflammatory bowel reaction with bowel wall thickening or fibrosis occurs,  
reduces in a dose dependent manner the fibrotic reaction to the inflammatory  
stimulus (Table 1). It is to be expected that in a similar manner NAG will prevent the  
development of fibrosis of the peritoneum in CAPD patients.

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In addition to glucose CAPD solutions typically also contain a suitable number and quantity of electrolytes such that a more less physiologic solution is obtained. For example, lactate is included as a base substitute. Its absorption and metabolism will correct metabolic acidosis. Sodium is usually included at a concentration slightly lower to that found in plasma, or 132–137 mM/L, to promote sodium removal. Similarly, chloride is usually included in the CAPD solution at physiologic strengths of 100–110 mM/L. .

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The normal osmolarity of blood is approximately 280 mOsm/L, so that a CAPD solution must have a greater osmotic value than this if it to be effective as a dialysis solution, and preferably it should have an osmotic pressure of 300–700 mOsm/L, and more specifically 310–560, or in a more limited range, of 350 to 450 mOsm/L (from Patent 4,879,280).

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**Table 1**

<b>COLON FIBROSIS (AS MEASURED BY WEIGHT(gm) OF 8 cm OF COLON)</b>	
<b>INTRARECTAL ADMINISTRATION</b>	<b>MEAN <math>\pm</math> SEM</b>
Control (20 mg TNB* in 0.25 mL Ethanol)	2.301 $\pm$ 0.222
25 mg NAG/kg BWt 1 hr before TNB/EtOH	1.669 $\pm$ 0.142
50 mg NAG/kg BWt 1 hr before TNB/EtOH	1.339 $\pm$ 0.155
100 mg NAG/kg BWt 1 hr before TNB/EtOH	1.150 $\pm$ 0.068

\* TNB = trinitrobenzenesulfonic acid

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In experiments in which rats were dialyzed for 4 hours with Hanks Balances salt solution with either glucose or N–acetylglucosamine added at a concentration of 75 mM or 214 mM, at a pH of 7.35 – 7.4. The net ultrafiltration was calculated as the difference between the drained volume of dialysate after 4 hours dwell time in the peritoneal cavity and the infused volume (20 mL) of the dialysis fluid. As well, the concentration of urea and creatinine in the blood and the dialysis fluid were measured. Permeability of the peritoneal membrane to urea and creatinine, expressed as the Mass Transfer Area Coefficient which was calculated according to the method of Krediet et al (Blood Purif 4: 194, 1986). The results, given in the Table below, clearly demonstrate that NAG results in a statistically significant increase in net ultrafiltration as well as peritoneal clearance of urea without increasing albumin or total protein loss into the dialysis fluid. In

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5 addition, the inclusion of NAG in the dialysate fluid stimulated the synthesis of hyaluronic acid, as shown by the more than 100% increase in amount of hyaluronic acid secreted in the dialysis fluid compared to the glucose treated rats. These in vivo experiments clearly demonstrate that NAG is a more effective osmotic agent than glucose when used for peritoneal dialysis.

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	<b>Glucose 75 mM (N=11)</b>	<b>NAG 75 mM (N=14)</b>	<b>Glucose 214 mM (N=11)</b>	<b>NAG 214 mM (N=13)</b>
<b>Net Ultrafiltration (mL/4 hrs)</b>	-0.44 ± 2.0	-0.11 ± 1.6	11.45 ± 1.2	<b>14.45 ± 1.6*</b>
<b>Mass Transfer Area Coef for Urea (mL/min)</b>	0.344 ± 0.13	0.287 ± 0.13	0.212 ± 0.07	0.262 ± 0.15
<b>Peritoneal Clearance of Urea (mL/min)</b>	18.8 ± 2.2	18.4 ± 2.1	26.9 ± 2.0	<b>30.0 ± 2.2**</b>
<b>Total Protein Dialysate/Serum Ratio (%)</b>	4.3 ± 1.0	4.4 ± 0.6	2.8 ± 0.4	3.1 ± 0.5
<b>Albumin Dilaysate/Serum Ratio (%)</b>	4.0 ± 1.6	3.9 ± 1.2	1.6 ± 0.6	2.0 ± 0.9
<b>Hyaluronic Acid in Dialysate Fluid (ug/L)</b>	103 ± 21	<b>226 ± 93*</b>	91 ± 31	<b>217 ± 96***</b>

\* = statistically significant ('t'-test), p < 0.001

\*\* = statistically significant ('test'-test), p < 0.01

\*\*\* = statistically significant, P < 0.002

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The stimulation of hyaluronic acid by N-acetylglucosamine was confirmed in tissue culture of human mesothelial cells.

20 As many changes can be made to the embodiments of the invention without departing from the scope of the invention, it is intended that all material herein be interpreted as illustrative of the invention and not in a limiting sense.